

Supporting Information for  
The Impact of Unimolecular Reactions on Acyl  
Peroxy Radical Initiated Isoprene Oxidation

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**Other calculated unimolecular reaction rate coefficients**

Table S1 presents the rest of the calculated unimolecular reaction rates for the acyl peroxy radicals studied. These reactions are also illustrated in Figure S1. For acr-APR, cyc3-APR and cyc4-APR 1,4 H-shifts, the reaction product decomposed similarly to ace-APR 1,4 H-shift. This decomposition led to the formation of a hydroperoxy radical and a ketene corresponding to the reacting APR structure.

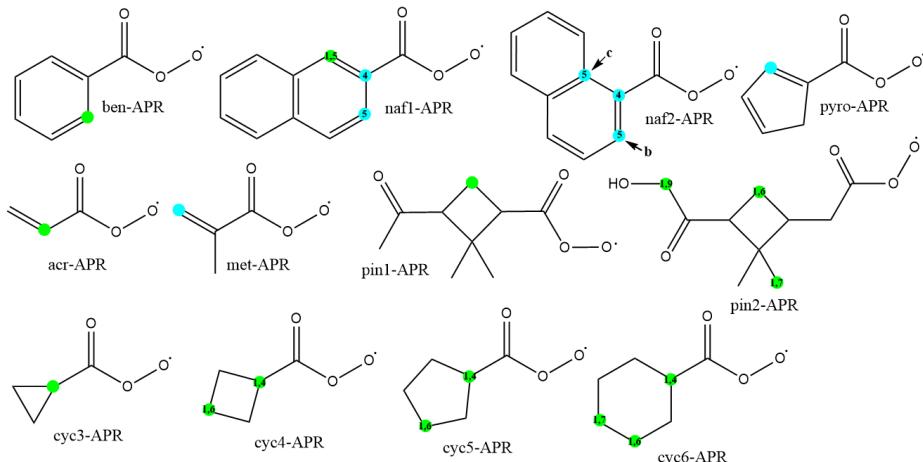


Figure S1: APR structures and studied unimolecular reactions corresponding to the results in Table S1. Reactions are marked with dots, green for H-shifts and blue for endoperoxide ring formations. For clarification, the prefixes for structures with multiple calculated rates are also marked.

Table S1: Calculated energy barrier heights ( $\Delta E^{\text{TS}}$  in kcal/mol), Eckart tunneling coefficients ( $\kappa_t$ ) for H-shifts and unimolecular MC-TST reaction rate coefficients ( $k_{\text{uni}}$  in  $\text{s}^{-1}$ ) at 298 K of other unimolecular reactions for the studied APRs.

| Radical  | Reaction                    | $\Delta E^{\text{TS}}$ | $\kappa_t$ | $k_{\text{uni}}$       |
|----------|-----------------------------|------------------------|------------|------------------------|
| Acr-APR  | 1,4 H-shift <sup>a</sup>    | 42.34                  | 2          | $1.75 \times 10^{-18}$ |
| Met-APR  | 5-endoperoxide              | 21.58                  | -          | $2.71 \times 10^{-4}$  |
| Ben-APR  | 1,5 H-shift                 | 29.76                  | 9          | $2.58 \times 10^{-9}$  |
| Pyro-APR | 5-endoperoxide              | 23.18                  | -          | $9.30 \times 10^{-6}$  |
| Cyc6-APR | 1,4 H-shift                 | 22.86                  | 42         | $4.79 \times 10^{-3}$  |
|          | 1,6 H-shift                 | 24.51                  | 32         | $6.19 \times 10^{-5}$  |
|          | 1,7 H-shift                 | 27.12                  | 32         | $4.36 \times 10^{-7}$  |
| Cyc5-APR | 1,4 H-shift                 | 22.87                  | 40         | $1.73 \times 10^{-3}$  |
|          | 1,6 H-shift                 | 22.22                  | 91         | $1.81 \times 10^{-3}$  |
| Cyc4-APR | 1,4 H-shift <sup>a</sup>    | 38.23                  | 3          | $9.63 \times 10^{-15}$ |
|          | 1,6 H-shift                 | 27.48                  | 238        | $1.24 \times 10^{-6}$  |
| Cyc3-APR | 1,4 H-shift <sup>a</sup>    | 39.72                  | 2          | $1.81 \times 10^{-16}$ |
| Pin1-APR | 1,5 H-shift                 | 24.42                  | 26         | $5.52 \times 10^{-5}$  |
| Pin2-APR | 1,6 H-shift                 | 22.88                  | 66         | $9.26 \times 10^{-4}$  |
|          | 1,7 H-shift                 | 22.79                  | 163        | $1.35 \times 10^{-3}$  |
|          | 1,9 H-shift                 | 24.96                  | 101        | $6.03 \times 10^{-6}$  |
| Naf1-APR | 1,5 H-shift                 | 29.57                  | 7          | $1.67 \times 10^{-9}$  |
|          | 4-endoperoxide              | 23.59                  | -          | $6.81 \times 10^{-6}$  |
|          | 5-endoperoxide              | 22.55                  | -          | $1.73 \times 10^{-5}$  |
| Naf2-APR | 4-endoperoxide              | 19.64                  | -          | $5.59 \times 10^{-3}$  |
|          | 5-endoperoxide <sup>b</sup> | 19.94                  | -          | $1.53 \times 10^{-3}$  |
|          | 5-endoperoxide <sup>c</sup> | 28.89                  | -          | $3.03 \times 10^{-10}$ |

<sup>a</sup> Reaction product decomposes. <sup>b&c</sup> See Figure S1.

## Comparison of $\omega$ B97X-D and M06-2X functionals for calculating rates for reactions between isoprene and APRs

Table S2 presents the calculated bimolecular reaction rate coefficients for the reactions of isoprene with ace-, pro- and ben-APR with two different functionals,  $\omega$ B97X-D and M06-2X, using MC-TST. Also, the LC-TST rates calculated using the  $\omega$ B97X-D functional are included.

Table S2: Calculated energy barrier heights ( $\Delta E^{\text{TS}}$  in kcal/mol) and bimolecular reaction rate coefficients ( $k_{\text{bi}}$  in  $\text{cm}^3\text{s}^{-1}$ ) at 298 K for ace-APR R4, pro-APR and ben-APR R1 reactions with isoprene calculated at the DLPNO-CCSD(T)/aug-cc-pVTZ// $\omega$ B97X-D/6-31+G\* (LC-TST and MC-TST) and DLPNO-CCSD(T)/aug-cc-pVTZ//M06-2X/6-31+G\* level (MC-TST).

| Radical | Reaction | Functional      | Theory | $\Delta E^{\text{TS}}$ | $k_{\text{bi}}$       |
|---------|----------|-----------------|--------|------------------------|-----------------------|
| Ace-APR | R4       | $\omega$ B97X-D | MC-TST | 2.8                    | $1.1 \times 10^{-17}$ |
|         |          | $\omega$ B97X-D | LC-TST |                        | $1.3 \times 10^{-17}$ |
|         |          | M06-2X          | MC-TST | 2.8                    | $6.0 \times 10^{-18}$ |
| Pro-APR | R1       | $\omega$ B97X-D | MC-TST | 1.3                    | $1.3 \times 10^{-16}$ |
|         |          | $\omega$ B97X-D | LC-TST |                        | $8.5 \times 10^{-17}$ |
|         |          | M06-2X          | MC-TST | 1.4                    | $2.8 \times 10^{-17}$ |
| Ben-APR | R1       | $\omega$ B97X-D | MC-TST | 0.1                    | $1.6 \times 10^{-15}$ |
|         |          | $\omega$ B97X-D | LC-TST |                        | $9.1 \times 10^{-16}$ |
|         |          | M06-2X          | MC-TST | 0.3                    | $4.9 \times 10^{-16}$ |

## PES graphs for reactions R1 and R4 between isoprene and OH

The PES graphs for the reactions R1 and R4 between isoprene and OH calculated with two different functionals,  $\omega$ B97X-D and M06-2X, are presented in Figures S2 and S3.

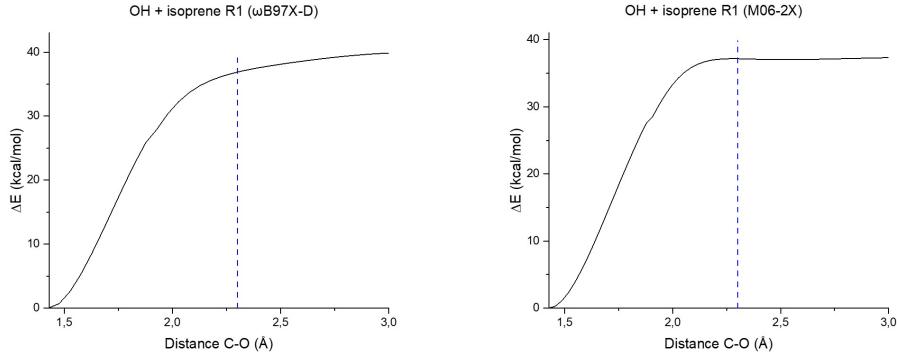


Figure S2: PES graphs for reaction R1 between isoprene and OH calculated with  $\omega$ B97X-D and M06-2X functionals. Energies are relative to the equilibrium structure. Blue dashed line shows C–O bond distance for TS structure optimized at the M06-2X/6-31+G\* level.

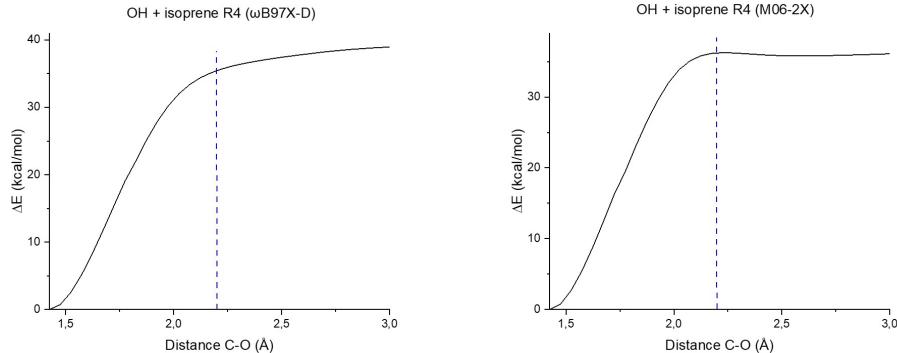


Figure S3: PES graphs for reaction R4 between isoprene and OH calculated with  $\omega$ B97X-D and M06-2X functionals. Energies are relative to the equilibrium structure. Blue dashed line shows C–O bond distance for TS structure optimized at the M06-2X/6-31+G\* level.

## Energy diagrams for ace-APR and OH reactions R1 with isoprene

Energy diagrams are presented in Figure S4. Energies for reactants, complex, TS and product were calculated at the DLPNO-CCSD(T)/aug-cc-pVTZ// $\omega$ B97X-D/6-31+G\* level for ace-APR and DLPNO-CCSD(T)/aug-cc-pVTZ//M06-2X/6-31+G\* level for OH.

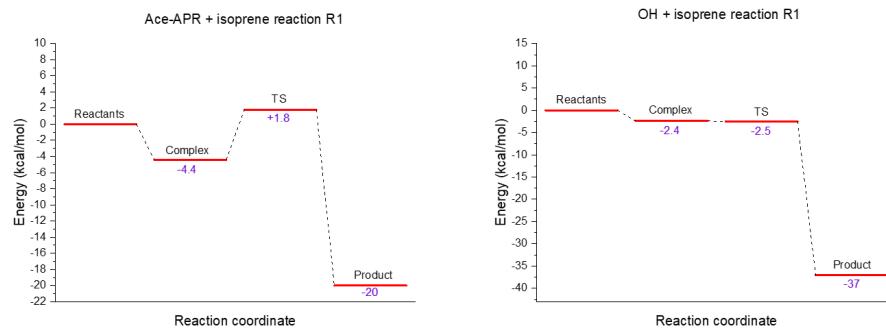


Figure S4: Energy diagrams for ace-APR and OH reactions R1 with isoprene calculated at the DLPNO-CCSD(T)/aug-cc-pVTZ// $\omega$ B97X-D/6-31+G\* and DLPNO-CCSD(T)/aug-cc-pVTZ//M06-2X/6-31+G\* level, respectively.